

## Fundamental Physical Constants — Frequently used constants

Quantity	Symbol	Value	Unit	Relative std. uncert. $u_r$
speed of light in vacuum	$c, c_0$	299 792 458	$\text{m s}^{-1}$	(exact)
magnetic constant	$\mu_0$	$4\pi \times 10^{-7}$ $= 12.566 370 614\dots \times 10^{-7}$	$\text{N A}^{-2}$ $\text{N A}^{-2}$	
electric constant $1/\mu_0 c^2$	$\epsilon_0$	$8.854 187 817\dots \times 10^{-12}$	$\text{F m}^{-1}$	(exact)
Newtonian constant of gravitation	$G$	$6.674 28(67) \times 10^{-11}$	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$	$1.0 \times 10^{-4}$
Planck constant	$h$	$6.626 068 96(33) \times 10^{-34}$	$\text{J s}$	$5.0 \times 10^{-8}$
$h/2\pi$	$\hbar$	$1.054 571 628(53) \times 10^{-34}$	$\text{J s}$	$5.0 \times 10^{-8}$
elementary charge	$e$	$1.602 176 487(40) \times 10^{-19}$	$\text{C}$	$2.5 \times 10^{-8}$
magnetic flux quantum $h/2e$	$\Phi_0$	$2.067 833 667(52) \times 10^{-15}$	$\text{Wb}$	$2.5 \times 10^{-8}$
conductance quantum $2e^2/h$	$G_0$	$7.748 091 7004(53) \times 10^{-5}$	$\text{S}$	$6.8 \times 10^{-10}$
electron mass	$m_e$	$9.109 382 15(45) \times 10^{-31}$	$\text{kg}$	$5.0 \times 10^{-8}$
proton mass	$m_p$	$1.672 621 637(83) \times 10^{-27}$	$\text{kg}$	$5.0 \times 10^{-8}$
proton-electron mass ratio	$m_p/m_e$	1836.152 672 47(80)		$4.3 \times 10^{-10}$
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	$\alpha$	$7.297 352 5376(50) \times 10^{-3}$		$6.8 \times 10^{-10}$
inverse fine-structure constant	$\alpha^{-1}$	137.035 999 679(94)		$6.8 \times 10^{-10}$
Rydberg constant $\alpha^2 m_e c / 2h$	$R_\infty$	10 973 731.568 527(73)	$\text{m}^{-1}$	$6.6 \times 10^{-12}$
Avogadro constant	$N_A, L$	$6.022 141 79(30) \times 10^{23}$	$\text{mol}^{-1}$	$5.0 \times 10^{-8}$
Faraday constant $N_A e$	$F$	96 485.3399(24)	$\text{C mol}^{-1}$	$2.5 \times 10^{-8}$
molar gas constant	$R$	8.314 472(15)	$\text{J mol}^{-1} \text{K}^{-1}$	$1.7 \times 10^{-6}$
Boltzmann constant $R/N_A$	$k$	$1.380 6504(24) \times 10^{-23}$	$\text{J K}^{-1}$	$1.7 \times 10^{-6}$
Stefan-Boltzmann constant $(\pi^2/60)k^4/\hbar^3c^2$	$\sigma$	$5.670 400(40) \times 10^{-8}$	$\text{W m}^{-2} \text{K}^{-4}$	$7.0 \times 10^{-6}$
electron volt: $(e/C) J$	Non-SI units accepted for use with the SI			
(unified) atomic mass unit	$eV$	$1.602 176 487(40) \times 10^{-19}$	$\text{J}$	$2.5 \times 10^{-8}$
$1 \text{ u} = m_u = \frac{1}{12} m(^{12}\text{C})$ $= 10^{-3} \text{ kg mol}^{-1}/N_A$	$u$	$1.660 538 782(83) \times 10^{-27}$	$\text{kg}$	$5.0 \times 10^{-8}$

P. J. Mohr, B. N. Taylor, and D. B. Newell (2007),  
 "The 2006 CODATA Recommended Values of the Fundamental Physical Constants"  
 (Web Version 5.0). This database was developed by J. Baker, M. Douma, and S. Kotchigova.  
 Available: <http://physics.nist.gov/constants>  
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